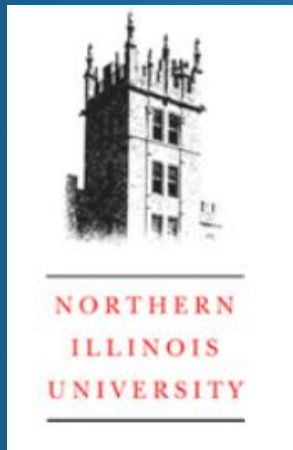


Hyper-alkaline Waters in Calumet Wetlands (South Chicago, IL)



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Introduction

- Hyper-alkaline (pH up to 13) surface and groundwater habitats
- Seasonal and site-specific differences in pH (9-13) and heavy metal concentration
- Unique site for study of alkaliphilic and alkalitolerant organisms





Chicago

20 km

**Calumet
area**



Industrial History of Calumet

- 1870s: building of Calumet Harbor – large scale industrial development
- Steel production- ore from Michigan
- Dumping of waste products to infill wetlands
 - Creating new land – very heterogeneous
 - Eliminating breeding grounds for insects





Big Marsh Site Grouping



Calumet Wetland & Wolf Lake

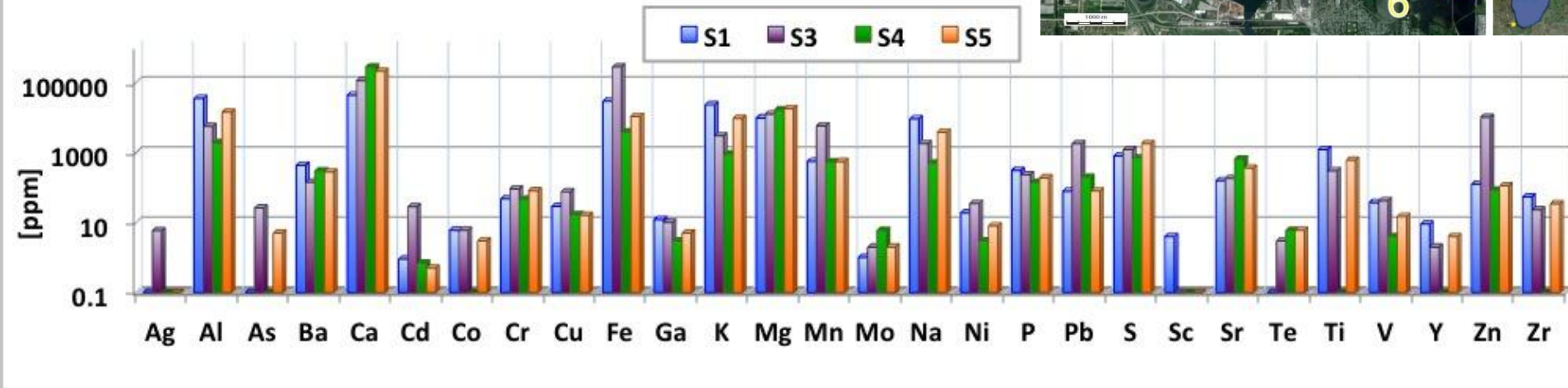
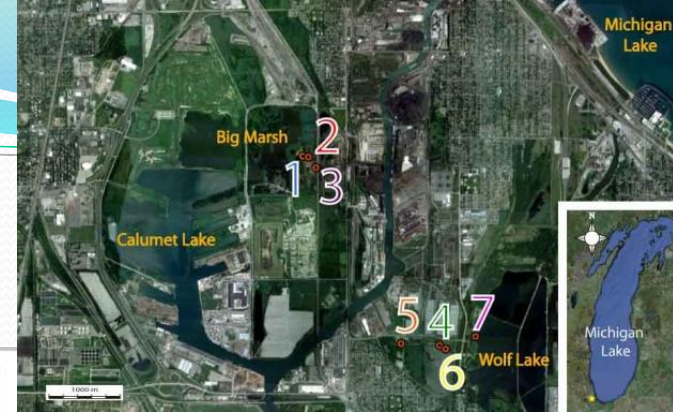


Geochemistry of Sites

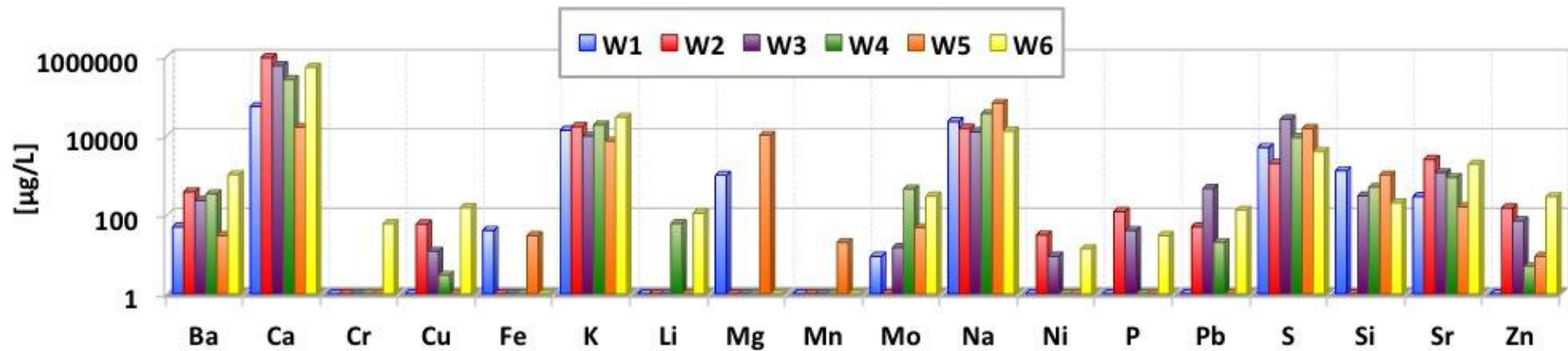
- High temperature slags with Ca-Si minerals
 - Includes metallic Fe and Mn and other steel additives
- Long term weathering:
 - Rankinite: $\text{Ca}_3\text{Si}_2\text{O}_7 + 7\text{H}_2\text{O} = 3\text{Ca}^{2+} + 2\text{H}_4\text{SiO}_4 + 6\text{OH}^-$
 - Larnite: $\text{Ca}_2\text{SiO}_4 + 4\text{H}_2\text{O} = 2\text{Ca}^{2+} + \text{H}_4\text{SiO}_4 + 4\text{OH}^-$
 - Akermanite:
 - $\text{Ca}_2\text{MgSi}_2\text{O}_7 + 7\text{H}_2\text{O} = 2\text{Ca}^{2+} + \text{Mg}^{2+} + 2\text{H}_4\text{SiO}_4 + 6\text{OH}^-$
- Origin of Ca-OH water with strongly temperature-dependent high pH
- On contact with atmosphere:
$$\begin{aligned}\text{CO}_2 + \text{H}_2\text{O} &= 2\text{H}^+ + \text{CO}_3^{2-} \\ \text{Ca}^{2+} + \text{CO}_3^{2-} &= \text{CaCO}_3\end{aligned}$$

Geochemistry

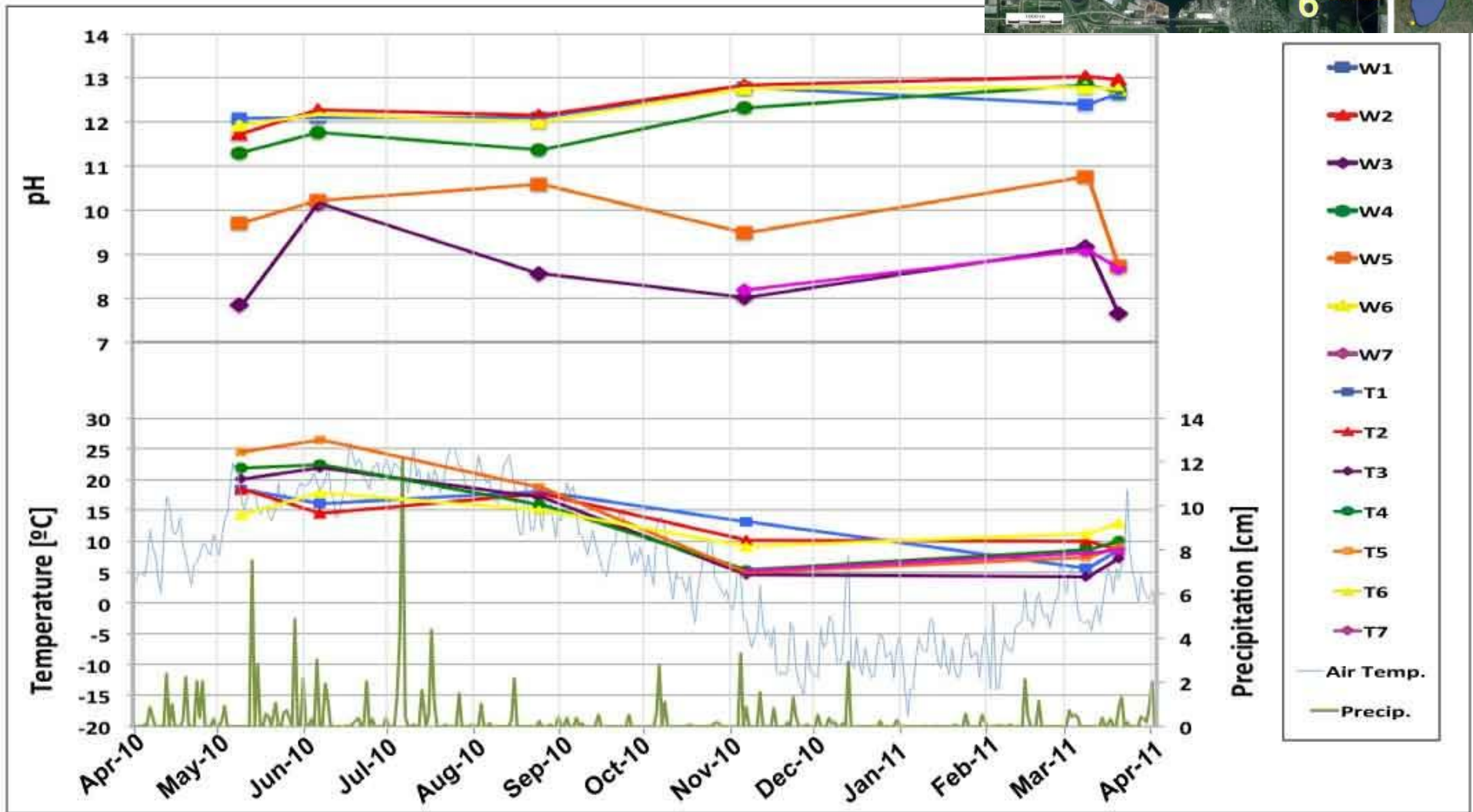
Soil



Water

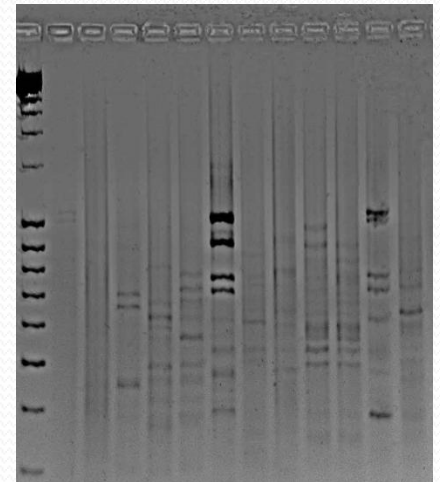


This aerial map illustrates the Calumet River watershed in Indiana, highlighting seven sampling locations (1-7) for the study. The map shows the river's course and surrounding land use, including agricultural fields and urban areas. Key features labeled include Big Marsh, Calumet Lake, Wolf Lake, and Michigan Lake. An inset map in the bottom right corner provides a broader geographical context, showing the study area's location within the state of Indiana.

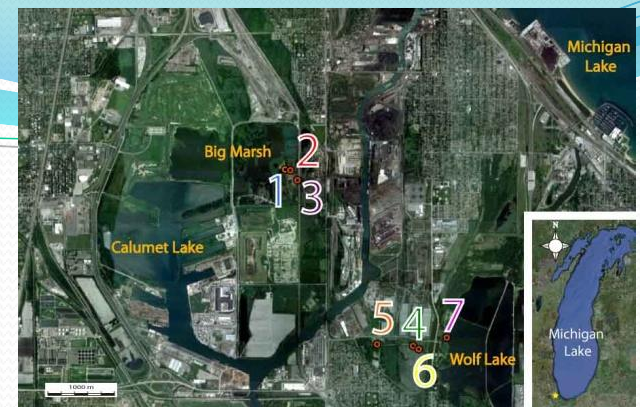


Microbial Community

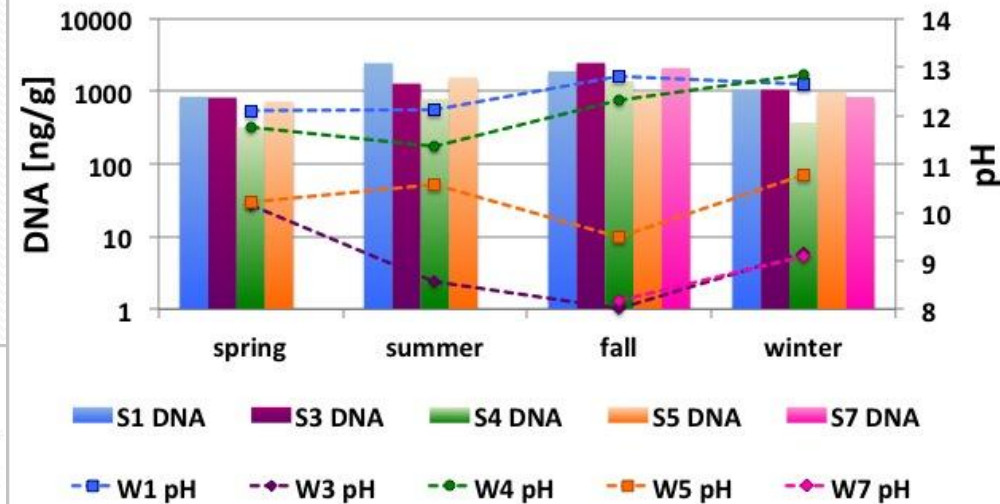
- From sediment and water, different seasons
 - Issues with methods in this environment
- DNA extracted using MoBio PowerSoil Isolation kit
- DNA was assessed using the Ribosomal Intergenic Spacer (RIS) primer set: Fail Safe buffer
 - 16S 1406F: TGY ACA CAC CGC CCG T
 - 23S 115R: GGG TTB CCC CAT TCR G
 - (Cardinale et al., 2004)
- Visualized on 2% SFR agarose TAE gel
 - Analyzed using GelQuest software (SequentiX)
 - DNA similarity index was calculated using a modified Jaccard index (Nei and Li, 1979)



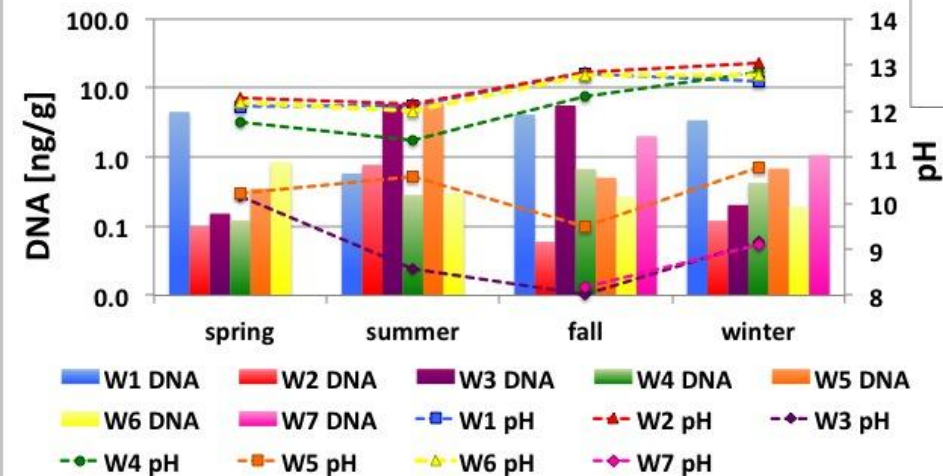
pH & DNA Conc.

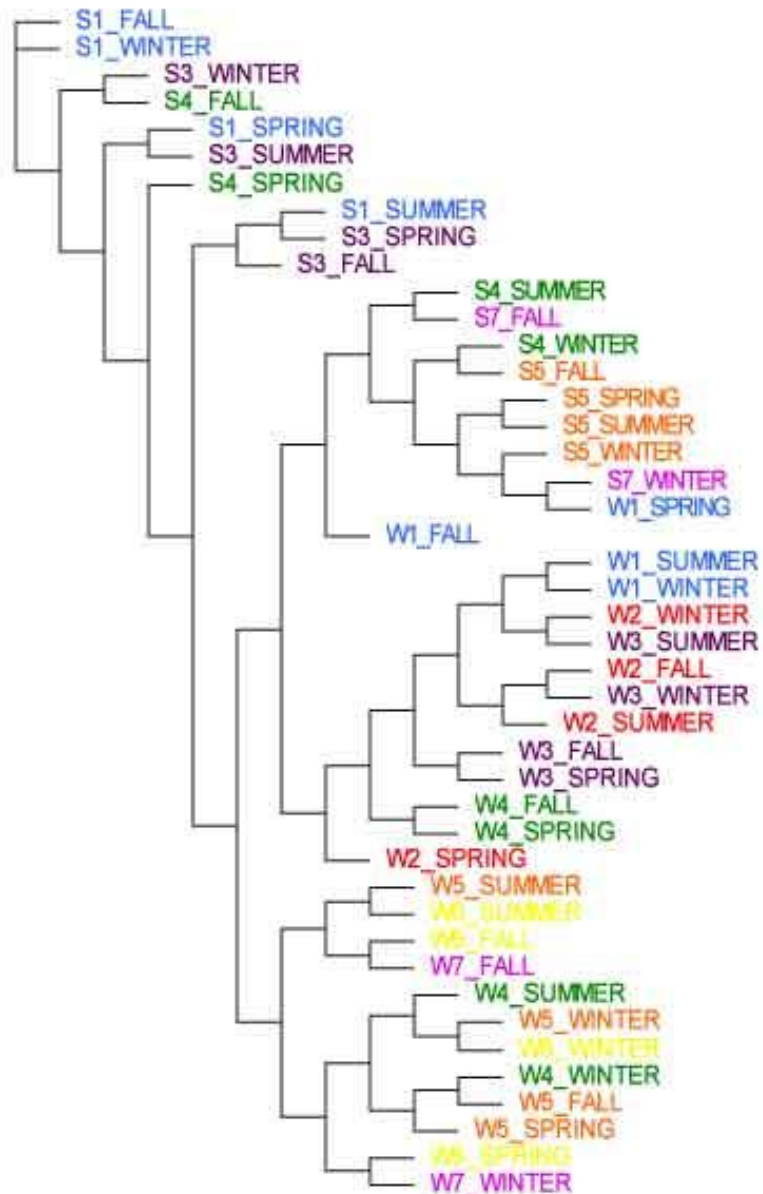


DNA & pH



DNA & pH



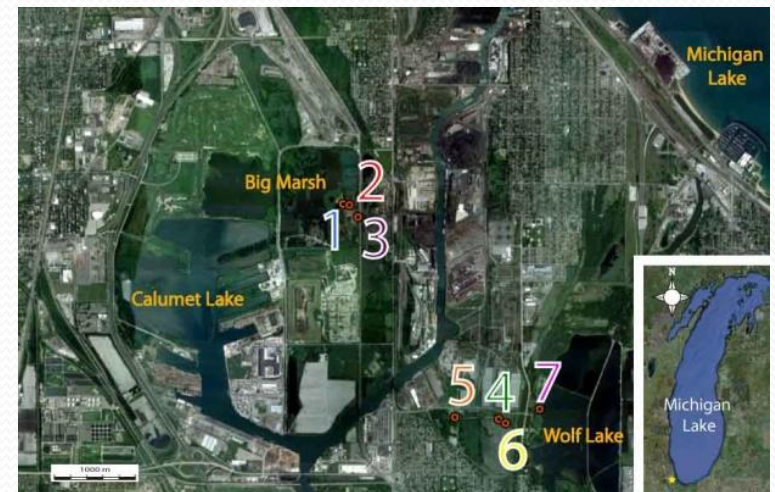
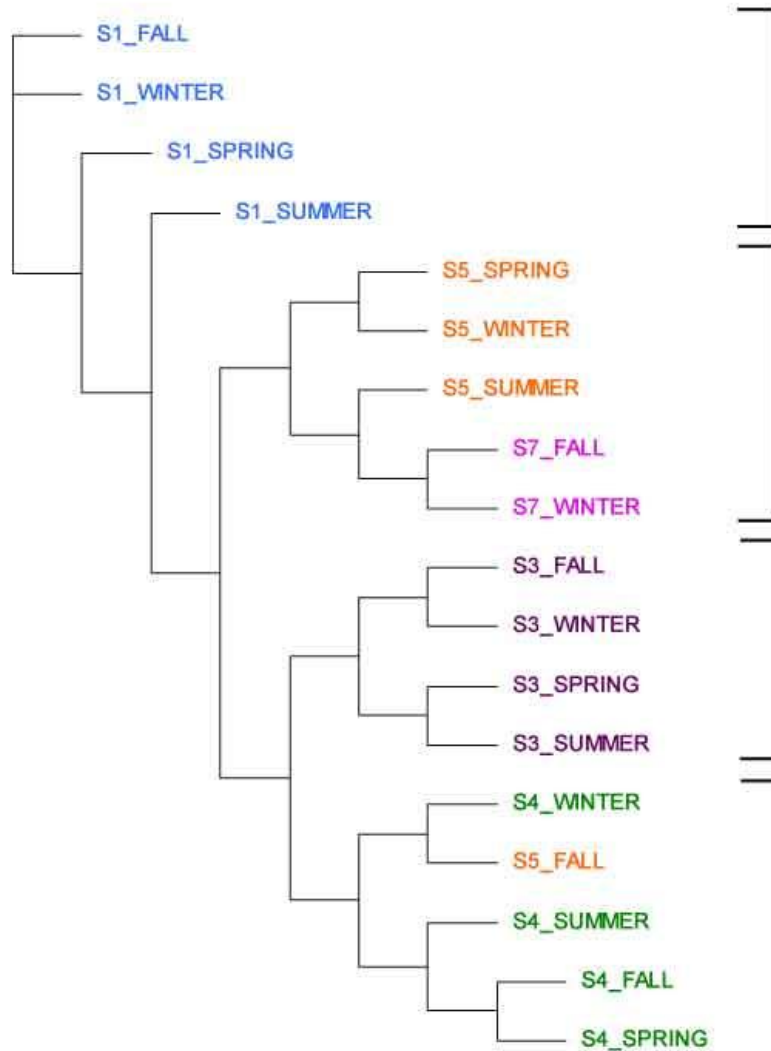


Soil

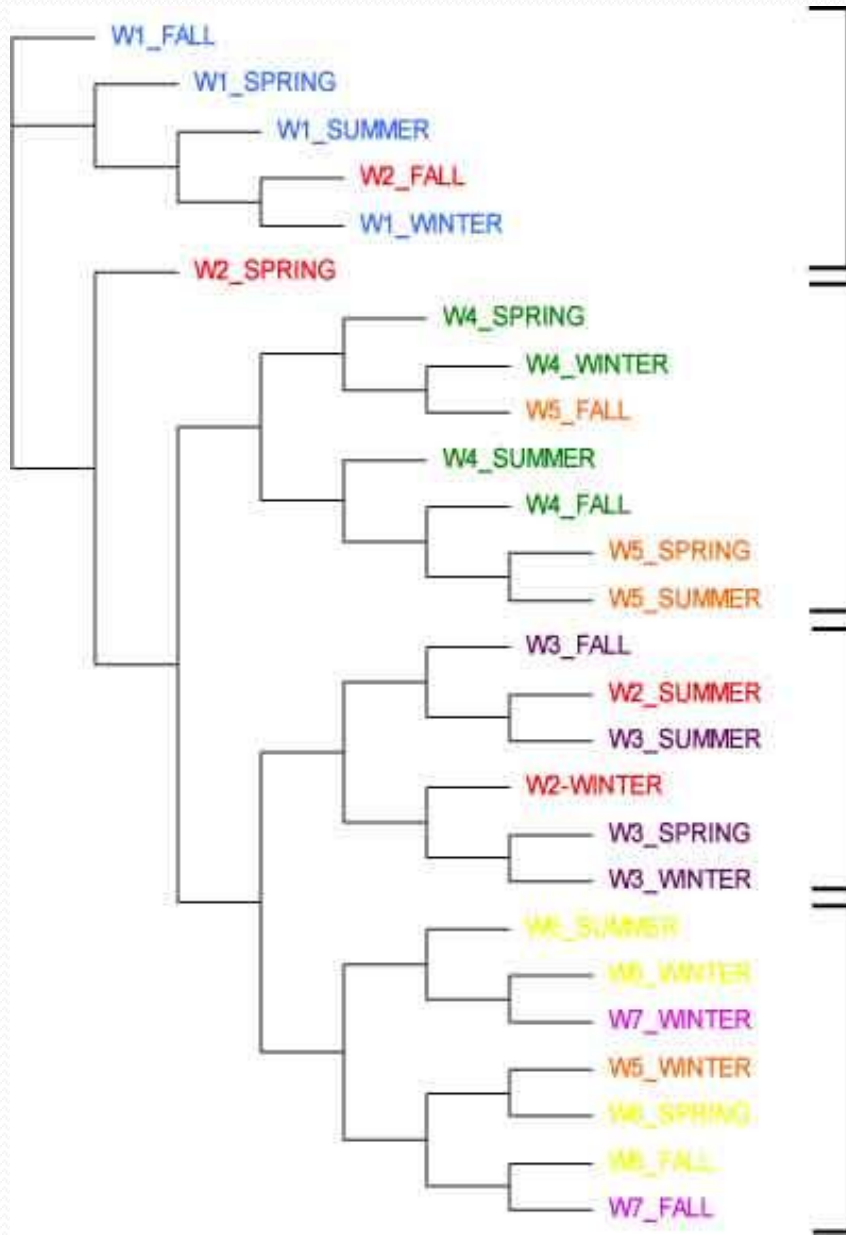


Water

Soil



Water



Remediation

- Laboratory columns are being constructed to test reactive barriers and determine how these influence the microbial communities.
- Sand, dolomite, and apatite II™ (Chen et al., 1997)
- Water and sediment from different sites
- Beginning this summer

Conclusions



- Chemical analysis of major ions varied that sites 1-6
- Seasonal monitoring and sampling revealed strong influence of temperature on pH at sites 1, 2, 4, and 6
 - Influence of pH on DNA concentrations was not significant
- DNA concentrations in soil samples were up to three orders magnitude higher than those in water samples
 - Soils providing microorganisms with microniche-protection throughout the year.
 - More research
- DNA fingerprint analysis showed:
 - Microbial communities differ between water and soil
 - Geographically related communities clustered closer together
 - More research

Acknowledgements

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