Using Isotopes to Understand Coupled Hydrological & Biological Processes

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Introduction

Understanding hydroecological connections is difficult! Isotopes can help! Discuss isotope examples from two main areas

- Radioactive Isotopes
 - Focus on water residence
 times
- Stable Isotopes
 - Focus on soil & the unsaturated zone





Radioactive Isotope Applications to Hydroecology

- Radioisotopes (e.g., carbon-14 and tritium) used to "label" ecohydrological processes
- Radioisotopes are underappreciated for understanding cycle or residence times

 e.g., groundwater residence times in aquifers can be long affecting biogeochemical processes at the river basin scale for many years.

• Examine water residence times of the upper Danube using tritium measurements of precipitation and river flow





Precipitation & Danube Tritium (Vienna, 1960-2008)



Time evolution of the T_P/T_R ratio in the Upper Danube





Lumped Parameter Modelling



Tritium in Vienna Precipitation (1960-2008)





Danube Tritium (Vienna, 1960-2008)





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Modelled Danube Tritium (Lumpy)





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Model Results

A two-box model provided the best fit.

 a short residence time component (piston flow assumption) and a longer residence time component (exponential flow assumption)

 60% "short" residence time component (mean residence time of 0.75 years)

 40% "long" residence time" component (mean residence time of 15 years)

•Average combined residence time is about 6 years.





Implications of Model Results

What do these results mean ecologically?

 Changes to biogeochemical and pollutant sources may take "on average" 6 years to become apparent in Vienna river water & over a decade to substantially decrease

• Estimations of residence/ cycle times are crucial at the basin scale & radio-isotopes are invaluable for this





Stable Isotopes (deuterium & oxygen-18)

 Stable isotopes are increasingly used to understand processes involving water along the plant/soil/atmosphere continuum

 In agriculture they are being used to improve water use efficiency

 Understand plant/water interactions in natural landscapes (ecohydrology)

 e.g., evaporation & evapotranspiration partitioning

- where plants get their water
- deep percolation
- climate variability





Where do plants get their water?

- Trees next to streams get their water from the stream right?
 - Wrong, they get it from groundwater!
 - Dawson & Ehleringer, 1991, Nature
 - Stable isotopes can be used to define where plants get their water



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FIG. 2 The hydrogen isotope ratio (δ D in %) obtained from xylem sap of three common streamside (closed symbols) and adjacent non-streamside (open symbols) tree species in the Red Butte Canyon Research Natural Area in 1989 as a function of the diameter at breast height (DBH) of the main trunk. Mean δ D values of stream water (from Fig. 1) and local well water (see text) were $-121.4 \pm 0.7\%$ and $-132.3 \pm 2.6\%$ respectively. Methods are outlined in the text and in ref. 10.

Vadose Zone Stable Isotopes



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IAEA

San Dimas, California Monocultures Newman & Graham, 2008, VZJ

Zimmerman et al. (1967, IAEA Symp. Series)

- A classic paper on vadose zone & ecohydrological stable isotope studies
- Percolation with tritium
- Evaporation & percolation with stable isotopes
 - Model for 2nd stage evaporation
- Plants (generally) don't fractionate soil water during root uptake - <u>a powerful concept!</u>

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Seasonal variation of deuterium content in the soil water of a grass-covered sandy soil near Heidelberg Solid line = samples from 0 - 10 cm depth Broken line = samples from 10 - 20 cm depth



FIG. 12. Deuterium content of soil water from two cores taken simultaneously from two neighbouring proving grounds on learny soil at Gießen. (Soil sampled 8 June, 1964.) The solid-line histogram tefers to the bare area, the broken line to the grass-covered one. The bare area contains on the average about 10% more deuterium

Isotope estimation of 2nd Stage Evaporation Rates (Allison & Barnes, 1983, Nature)



Chakir et al. Tunisian profile fit



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- Depth profiles of isotopes can quantify 2nd stage evaporation
- For bare soil or sparse vegetation
- Iteratively fit the bulge and steady state region using:
- $(\delta \delta_{res}) = (\delta_{ef} \delta_{res}) exp(-f(z)/\tilde{z}) \&$

• $\check{Z} = \theta_{ave} \tau D/E$

 Where δ are the delta values at a given depth, at the evaporation front and at depth respectively, f(z) depth fn for variable water content, ž is the decay length, θ is the ave. water content, and D is the diffusion coefficient of HDO in liquid water.

Evaporation estimates for multiple events from a single borehole (Chakir et al. Tunisia)

Phase	Thickness of the soil above the phase (cm)	Evaporation rate (mm/year)		Mean Evaporation rate (mm/year)
		τ=0.6	τ =0.67	
Α	50	20,7	23,2	22,0
В	175	15,6	17,4	16,5
С	280	10,2	11,4	10,8
D	345	11,0	12,3	11,7
E	200	6,4	7,2	6,8



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Vadose Zone Sampling for Isotopes (deuterium and oxygen-18)

- Direct sampling of vadose water in field
- Suction cups
- Zero tension samplers
- Passive wick lysimeters (flux and water samples)
- Passive wicks for snowmelt percolation (Frisbee et al., 2010, pts. 1 & 2, HP)
- No extraction necessary
- Difficult to sample deeper vadose zone





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Cumulative percolation at 1 meter depth using flux meters in a semiarid arroyo in New Mexico

 Multiple flux meters can be used to understand study site spatial variability

 New Mexico results show the importance of topography on percolation

for more information
see Gee et al., WRR
2008



Flux meters for stable isotope sampling

Cotton field percolation at 1 meter depth, Indus Basin, Pakistan

Isotope
 results show
 effects of
 transient
 percolation &
 subsequent
 evaporation
 events





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from Naveed Iqball

Vadose Zone Sampling for Isotopes (deuterium and oxygen-18)

"Indirect" sampling by collecting soil, sediments, or rock from the vadose zone

- Hand augering
- Coring with drill rig or direct push system
- Can sample very deep profiles
- Good storage potential
- Requires extraction





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Soil sample extraction for isotopes

- Cryogenic Vacuum Distillation
- Azeotropic Distillation
- Both methods are tedious & are often problematic (see e.g., Araguas-Araguas, 1995, J. of Hydr.)
- There may now be a better way!





Recent Development

- A better way for soil/core porewater isotope analyses?
- Equilibration method for soil samples using an laser stable isotope analyzer for water vapor (Wassenaar et al., 2008, EST)
 - In field, put sample in zip top bag, in lab fill with dry air, equilibrate
 - Using transfer line with needle, pump air directly into instrument
 - up to 30 samples/day

 This method is much less prone to problems than traditional extraction methods & is faster & easier





Example Laser Spectrum

Configure

 Each type of water molecule has a unique optical absorption



h2o 20060929 000.txt

27.21 °C

10:15:08

6.608 us

File Transfer Setup



Conclusions



- Isotopes can be a powerful approach for addressing a broad range of hydro-ecological problems
 - Evaporation and evapotranspiration partitioning
 - Deep percolation
 - Residence/cycle times using radioactive isotopes
 - Many others

Thank You!